

Alleima® Ti Grade 2

Tube and pipe, seamless

Datasheet

Alleima® Ti Grade 2 is a Commercially Pure (CP) titanium seamless tubing characterized by:

- Excellent resistance to general corrosion in seawater
- Resistant to stress corrosion cracking in chloride and sour gas environments
- Excellent resistance to pitting, crevice, and erosion corrosion
- High heat transfer efficiency
- Good formability and weldability
- Very low thermal expansion
- Moderate strength

Standards

- ASTM: Grade 2
- UNS: R50400

Chemical composition (nominal)

Chemical composition (nominal) %

N	H	O	Fe	C
≤0.03	≤0.015	≤0.25	0.30	0.08

Applications

Commercially Pure Ti Grade 2 titanium tubing provides excellent service in aggressive chloride-containing environments. The excellent mechanical, physical, and corrosion resistance properties of this grade make it an economical choice for many applications by reducing the product life cycle costs of equipment. Typical applications include:

Oil and gas industry

Chloride environments such as seawater handling and process systems and hydraulic and process fluid tubes in umbilicals.

Seawater cooling

Tubing for heat exchangers and coolers on oil platforms, in refineries, chemical industries, process industries, and other industries using seawater or chlorinated seawater as coolant.

Refineries and petrochemical plants

Heat exchangers and condensers where the process environment contains chlorides, sulfides, organics, organic acids, nitric acid, or wet chlorine.

Geothermal wells

Heat exchangers in geothermal exploitation units, systems exposed to geothermal or high-salinity brines, tubing, and casing for production.

Pulp and paper industry

Tubing for chloride-containing bleaching environments.

Desalination plants

Tube and pipe for seawater transport, heat exchanger tubing, and pressure vessels for reverse osmosis units.

Corrosion resistance

Titanium should not be used with strong reducing acids, fluoride solutions, pure oxygen, or anhydrous chlorine.

General corrosion

The general corrosion rates for Grade 2 titanium in a variety of media are shown in Table 1. CP titanium exhibits good corrosion resistance to a wide variety of environments including:

- Seawater and brines
- Inorganic salts
- Moist chlorine gas
- Alkaline solutions
- Oxidizing acids
- Organics and organic acids
- Sulfur compounds

Titanium corrosion rate data

Commercially pure grades

C=Concentration %
 T = Temperature °F (°C)
 R = Corrosion rate, mpy (mm/year)

Media	C	T	R	Media	C	T	R
Acetaldehyde	75	300 (149)	0.02(0.001)	Barium chloride	25	212 (100)	nil
	100	300 (149)	nil	Barium hydroxide	saturated	room	nil
Acetic acid	5 to 99.7	255 (124)	nil	Barium nitrate	10	room	nil
Acetic anhydride	99.5	boiling	0.5 (0.013)	Barium fluoride	saturated	room	nil
Acidic gases	-	100-500	<1.0 (<0.025)	Benzoic acid	saturated	room	nil
containing CO ₂ ,		(38-260)		Boric acid	saturated	room	nil
H ₂ O, Cl ₂ , SO ₂ , SO ₃ ,				Boric acid	10	boiling	nil
H ₂ O ₂ , NH ₃				Bromine	liquid	86 (30)	rapid)
Adipic acid	67	450 (232)	nil	Bromine moist	vapor	86 (30)	<0.1 (<0.003)
Aluminium chloride,				N-butylric acid	undiluted	room	nil
acrated	10	212 (100)	0.09 (0.002)*	Calcium bisulfite	cooking	79 (26)	0.02 (0.001)
Aluminium chloride,				liquor			
acrated	25	212 (100)	124 (3.15)*	Calcium carbonate	saturated	boiling	nil
Aluminium fluoride	saturated	room	nil	Calcium chloride	5	212 (100)	0.02 (0.005)*
Aluminium nitrate	saturated	room	nil	Calcium chloride	10	212 (100)	0.29 (0.007)*
Aluminium sulfate	saturated	room	nil	Calcium chloride	55	220 (104)	0.02 (0.001)*
Aluminium acid				Calcium hydroxide	saturated	boiling	nil
phosphate	10	room	nil	Calcium hypochlorite	6	212 (100)	0.05 (0.001)
Ammonia anhydrous	100	104 (40)	<5.0 (<0.127)	Calcium hypochlorite	18	70 (21)	nil
Ammonium acetate	10	room	nil	Calcium hypochlorite	saturated	-	nil
Ammonium				slurry			
bicarbonate	50	212 (100)	nil	Carbon dioxide	100	-	excellent
Ammonium bisulfite	spent	159 (71)	0.6 (0.015)	Carbon tetrachloride	vapor &	boiling	nil
pH 2.05	pulping			liquid			
liquor				Chlorine gas, wet	>0.7 H ₂ O	room	nil
Ammonium chloride	saturated	212 (100)	<0.5 (<0.013)	Chlorine gas, wet	>1.5 H ₂ O	392 (200)	nil
Ammonium hydroxide	28	room	0.1 (0.003)	Chlorine header sludge	-	207 (97)	0.03 (0.001)
Ammonium nitrate	28	boiling	nil	and wet chlorine			
Ammonium nitrate+	28	boiling	nil	Chlorine gas, dry	<0.5 H ₂ O	room	may react
1% nitric acid				Chlorine dioxide in	5	210 (99)	nil
Ammonium sulfate	10	212 (100)	nil	steam			
Aqua regia	3:1	room	nil	Chlorine trifluoride	100	<86 (30)	vigorous
Aqua regia	3:1	175 (79)	34.8 (0.884)				reaction

*May corrode in crevice

"Titanium, The Choice", Titanium Development Association, 1990.

Cont.

Media	C	T	R	Media	C	T	R
Chloroacetic acid	100	boiling	<5.0(<0.127)	Linseed oil, boiled	–	room	nil
Chlorosulfonic acid	100	room	7.5-12.3 (0.191-0.312)	Lithium chloride	50	300 (149)	nil*
Chloroform	vapor & liquid	boiling	0.01 (0.000)	Magnesium chloride	5-40	boiling	nil*
Chromic acid	10	boiling	0.1 (0.003)	Magnesium hydroxide	saturated	room	nil
Chromic acid	50	180 (82)	1.1 (0.028)	Magnesium sulfate	saturated	room	nil
Chromic acid + 5% nitric acid	5	70 (21)	<0.1 (<0.003)	Magnous chloride	5-20	212 (100)	nil
Citric acid	50	140 (60)	0.01 (0.000)	Maleic acid	18-20	95 (35)	0.6 (0.002)
Citric acid	50	212 (100)	<5.0(<0.127)	Mercuric chloride	saturated	212 (100)	<5 (<0.127)
Citric acid	50	boiling	5.50 (0.127-1.27)	Mercuric cyanide	saturated	room	nil
Cupric chloride	40	boiling	0.2 (0.005)	Methyl alcohol	91	95 (35)	nil
Cupric chloride	55	246 (119) (boiling)	0.1 (0.003)	Nickel chloride	20	212 (100)	0.11 (0.003)
Cupric cyanide	saturated	room	nil	Nitric acid, aerated	50	room	0.08 (0.002)
Cuprous chloride	50	194 (90)	<0.1 (<0.003)	Nitric acid, aerated	70	Room	0.18 (0.005)
Cyclohexane	–	302 (150)	0.1 (0.003)	Nitric acid, aerated	10	104 (40)	0.10 (0.003)
Dichloroacetic acid	100	boiling	0.29 (0.007)	Nitric acid, aerated	70	158 (70)	1.56 (0.040)
Dichlorobenzene + 4-5% HCl	–	355 (179)	4 (0.102)	Nitric acid, aerated	40	392 (200)	24 (0.610)
Diethylene triamine	100	room	nil	Nitric acid, aerated	20	554 (290)	12 (0.305)
Ethyl alcohol	95	boiling	0.5(0.013)	Nitric acid, non-aerated	70	176 (80)	1-3 (0.025-0.076)
Ethylene dichloride	100	boiling	0.2-5.0 (0.005-0.127)	Nitric acid	17	boiling	3-4 (0.076-0.102)
Ethylene diamine	100	room	nil	Nitric acid	35	boiling	5-20 (0.127-0.308)
Ferric chloride	10-20	room	nil	Nitric acid	70	boiling	2.5-37 (0.064-0.940)
Ferric chloride	10-50	boiling	nil	Nitric acid	–	room	0.1 (0.003)
Ferric chloride	50	302 (150)	0.1 (0.003)	Nitric acid white fuming	–	360 (160)	<5.0(<0.127)
Ferric sulfate	10	room	nil	Nitric acid red fuming	<about 2% H ₂ O	room	ignition sensitive
Fluoboric acid	5-20	elevated	rapid	Nitric acid	>about 3% H ₂ O	room	not ignition sensitive
Fluorosilicic	10	room	1870 (47.5)	Nitric acid + 10% NaClO	40	boiling	4.8-7.4 (0.122-0.188)
Food products	–	ambient	no attack	Oil well crudes	–	ambient	0.12-1.40 (0.003-0.036)
Formaldehyde	37	boiling	nil	Oxalic acid	1	boiling	0.26-23.2 (0.007-0.589)
Formic acid aerated	25	212 (100)	0.04(0.001)**	Oxalic acid	25	140 (60)	4247 (107.9)
Formic acid aerated	90	212 (100)	0.05 (0.001)**	Phenol	saturated solution	70 (21)	470 (11.9)
Formic acid non-aerated	25	212 (100)	96 (2.44)**	Phosphoric acid	10-30	room	4.0(0.102)
Formic acid non-aerated	90	212 (100)	118 (3.00)**	Phosphoric acid	30-80	room	0.8-2 (0.020-0.051)
Furfural	100	room	nil	Phosphoric acid	1	boiling	2-30 (0.051-0.762)
Glucosac acid	50	room	nil	Phosphoric acid + 3% nitric acid	81	190 (88)	10 (0.254)
Glycerin	–	room	nil	Phosphorous	100	room	15 (0.381)
Hydrogen chloride	dry gas	ambient	nil	Phosphorous oxychloride	100	room	0.14 (0.004)
Hydrochloric acid	1	boiling	>100 (>2.54)	Phosphorous trichloride	saturated	room	nil
Hydrochloric acid chlorine saturated	5	boiling	400 (10.2)	Pthalic acid	saturated	room	nil
+ 5% HNO ₃	5	280 (93)	<1 (<0.025)	Potassium bromide	saturated	room	nil
+ 5% HNO ₃	1	boiling	>1120 (>28.5)	Potassium chloride	saturated	room	nil
+ 0.5% CrO ₃	5	200 (93)	1.2(0.030)	Potassium ferricyanide	saturated	room	nil
+ 1% CrO ₃	5	100 (38)	0.72 (0.018)	Potassium hydroxide	50	80 (29)	0.4 (0.010)
+ 0.05% CuSO ₄	5	200 (93)	3.6 (0.091)	Potassium hydroxide	10	boiling	<5.0(<0.127)
+ 0.5% CuSO ₄	5	200 (93)	2.4 (0.061)	Potassium hydroxide	25	boiling	12 (0.305)
+ 0.5% CuSO ₄	5	boiling	3.3 (0.084)	Potassium sulfate	10	room	nil
Hydrofluoric acid	1.48	room	rapid	Potassium thiosulfate	1	–	nil
Hydrogen peroxide	6	room	<5 (<0.127)	Salicylic acid (Na salt)	saturated	room	nil
Hydrogen peroxide	30	room	<12 (<0.305)	Seawater	–	76 (24)	nil
Hydrogen sulfide	7.65, moist	200-230 (93-110)	nil	Sebacic acid	–	464 (240)	0.3 (0.008)
Hypochlorous acid + ClO ₂ and Cl ₂	17	100 (38)	0.001 (0.000)	Silver nitrate	50	room	nil
Iodine in water + Potassium Iodide	–	room	nil	Sodium acetate	saturated	room	nil
Lactic acid	10-85	212 (100)	<5.0(<0.127)				
Lead acetate	saturated	room	nil				

Cont.

Media	C	T	R	Media	C	T	R
Sodium aluminate	25	boiling	3.6 (0.091)	Sulfuric acid, aerated	3	140 (60)	0.5 (0.013)
Sodium bifluoride	saturated	room	rapid		5	140 (60)	190 (4.83)
Sodium bisulfate	saturated	room	nil		3	212 (100)	920 (23.4)
Sodium bisulfate	10	150 (66)	72 (1.83)		concentrated	room	62 (1.57)
Sodium chloride	23	boiling	nil*	Sulfuric acid	1	boiling	700 (17.8)
pH 1.5				Sulfuric acid	5	200 (93)	nil
Sodium chloride	23	boiling	28 (0.711)*	+0.25% CuSO ₄	30	100 (38)	2.4(0.061)
pH 1.2				+ 0.25% CrO ₃	30	200 (93)	nil
Sodium chloride	23	boiling	nil*	+ 1.0% CuSO ₄	30	boiling	65 (1.65)
pH 1.2 some dissolved chlorine				Sulfuric acid vapors	96	150 (66)	nil
Sodium citrate	saturated	room	nil	Sulfuric acid			
Sodium cyanide	saturated	room	nil	+ 10% HNO ₃	90	room	18 (0.457)
Sodium dichromate	saturated	room	nil	+ 70% HNO ₃	30	room	4.0(0.102)
Sodium fluoride	saturated	room	0.3 (0.008)	+ 90% HNO ₃	10	room	nil
Sodium bisulfite	25	boiling	nil	Sulfuric acid saturated	62	60 (16)	0.07 (0.002)
Sodium carbonate	25	boiling	nil	with chlorine			
Sodium chlorate	saturated	room	nil	Sulfuric acid saturated	5	374 (190)	<1 (<0.025)
Sodium hydroxide	5-30	70 (21)	>0.12 (>0.001)	with chlorine			
Sodium hydroxide	10	boiling	0.84 (0.021)	Sulfuric acid+	40	212 (100)	passive
Sodium hydroxide	40	176 (80)	5.0 (0.127)	4.79 g/l Ti ⁴⁺			
Sodium hydroxide	50	135 (57)	0.5(0.127)	Sulfurous acid	6	room	nil
Sodium hydroxide	73	265 (129)	7.0(0.178)	Tannic acid	25	212 (100)	nil
Sodium hydroxide	50-73	370 (188)	>43 (>1.09)	Tartaric acid	10-50	212 (100)	<5(<0.127)
Sodium hypochlorite	6	room	nil	Terephthalic acid	77	425 (218)	nil
Sodium nitrate	saturated	room	nil	Tetrachloroethalene,	100	boiling	0.02 (0.001)
Sodium phosphate	saturated	room	nil	liquid and vapor			
Sodium silicate	25	boiling	nil	Tetrachloroethylene +	-	boiling	5(0.127)
Sodium sulfate	10-20	boiling	nil	H ₂ O			
Sodium sulfide	saturated	room	nil	Tetrachloroethylene	100	boiling	nil
Sodium sulfite	saturated	boiling	nil	Titanium tetrachloride	99.8	572 (300)	62 (1.57)
Sodium thiosulfate	25	boiling	nil	Titanium tetrachloride	concentrated	room	nil
Soils, corrosive	-	ambient	nil	Trichloroacetic acid	100	boiling	573 (14.6)
Stannic chloride	24	boiling	1.76 (0.045)	Trichloroethylene	99	boiling	0.1-5 (0.003-0.127)
Stannic chloride	saturated	room	nil	Urea + 32% ammonia	28	360 (182)	3.1 (0.079)
Steam+ air	-	180 (82)	0.01 (0.000)	+ 20.5% H ₂ O, 19% CO ₂			
Succinic acid	100	365 (18.5)	nil	Water, degassed	-	600 (316)	nil
Sulfamic acid	3.75 g/l	boiling	nil	X-ray developer	-	room	nil
Sulfamic acid	7.5 g/l	boiling	108 (2.74)	solution			
Sulfamic acid +	7.5 g/l	boiling	1.2 (0.030)	Zinc chloride	20	220 (104)	nil*
375 g/FeCl ₃				Zinc chloride	50	302 (150)	nil*
Sulfur dioxide, water saturated	near 100	room	0.1 (0.003)	Zinc chloride	75	392 (200)	24 (0.610)**
Sulfur dioxide gas + small amount SO ₃ and approx 3% O ₂	18	600 (316)	0.2 (0.006)				

Crevice corrosion

CP titanium exhibits good resistance to crevice corrosion in salt solutions compared to stainless steels. CP titanium will not exhibit crevice corrosion at temperatures under 80 °C (176°F) regardless of pH, even under super chlorinated conditions.

Stress corrosion cracking

Grade 2 titanium shows excellent resistance to stress corrosion (SCC) cracking in hot chloride solutions and is immune to sec in seawater.

Erosion corrosion

Titanium shows excellent resistance to erosion in flowing seawater with velocities up to 130 ft/sec (40 m/sec) showing negligible effect on the material. The presence of abrasive particles, such as sand, has only a small effect on corrosion.

Hydrogen embrittlement

There is no significant absorption of hydrogen into titanium exposed to seawater, even at higher temperatures. Normally hydrogen absorption occurs only when the three following conditions are met:

1. pH is < 3 or > 12
2. Temperature is above 176F (80C)
3. A mechanism exists for hydrogen generation such as a galvanic couple or impressed current.

Bio-corrosion

Titanium alloys have demonstrated a unique immunity to all forms of microbiologically influenced corrosion. Since titanium alloys do not display any toxicity toward marine organisms, biofouling can occur in seawater. This can be minimized by chlorination or by increasing the water velocity through the heat exchanger.

Corrosion fatigue

Titanium, unlike many other materials, does not show a decrease in fatigue performance in the presence of seawater. Both fatigue endurance limits and fatigue crack growth rates are the same whether tested in air or seawater.

Galvanic corrosion

In the galvanic series, titanium is towards the noble end near stainless steels, and will normally act as the cathode when coupled with other metals. The titanium will therefore not be affected by galvanic corrosion but can accelerate corrosion of the other metal. Coupling of titanium with more noble metals, such as graphite, only enhances titanium's passivity.

Fabrication

Bending

Titanium tubing can be bent at room temperature using standard bend tooling and techniques. When bending

thin walled tubing or if a tight bend radius is needed, a mandrel should be used for adequate support of the ID. The mandrel should be well lubricated in order to prevent galling of the ID surface. Due to the moderate strength and low modulus of this alloy, springback is about twice that of stainless steel and must be taken into account.

Roller expansion

Titanium tubing can be roller expanded into tube sheets similar to other tubing materials. The suggested wall reduction for titanium is 10% to provide optimum pull out strength.

Machining and cutting

Machining and cutting titanium tubing is routine when the following procedures are used:

- Use low cutting speeds and high feed rates
- Use large volumes of coolant
- Use sharp tools and replace as soon as worn
- Never stop feeding while tool is in contact with workpiece

Tubing and pipe specifications

ASTM B337: Seamless and welded pipe

ASTM B338: Seamless and welded tubing

ASME SB338: Seamless and welded tubing

AMS 4943: Aerospace hydraulic tubing, annealed

AMS 4944/4945: Aerospace hydraulic tubing, cold worked and stress relieved

DIN/VD/TUV 230/2: Seamless tube and pipe

Approved by the American Society of Mechanical Engineers (ASME) Boiler and Pressure Vessel Code, Section VIII, Div. 1, Case 2081

Sizes and surface conditions

Tube and pipe are supplied in the cold reduced or cold reduced and annealed condition. Tubing can be delivered in the following surface conditions: as cold pilgered, acid etched, or belt polished. The principal size range for seamless products is shown as the white area in figure 1.

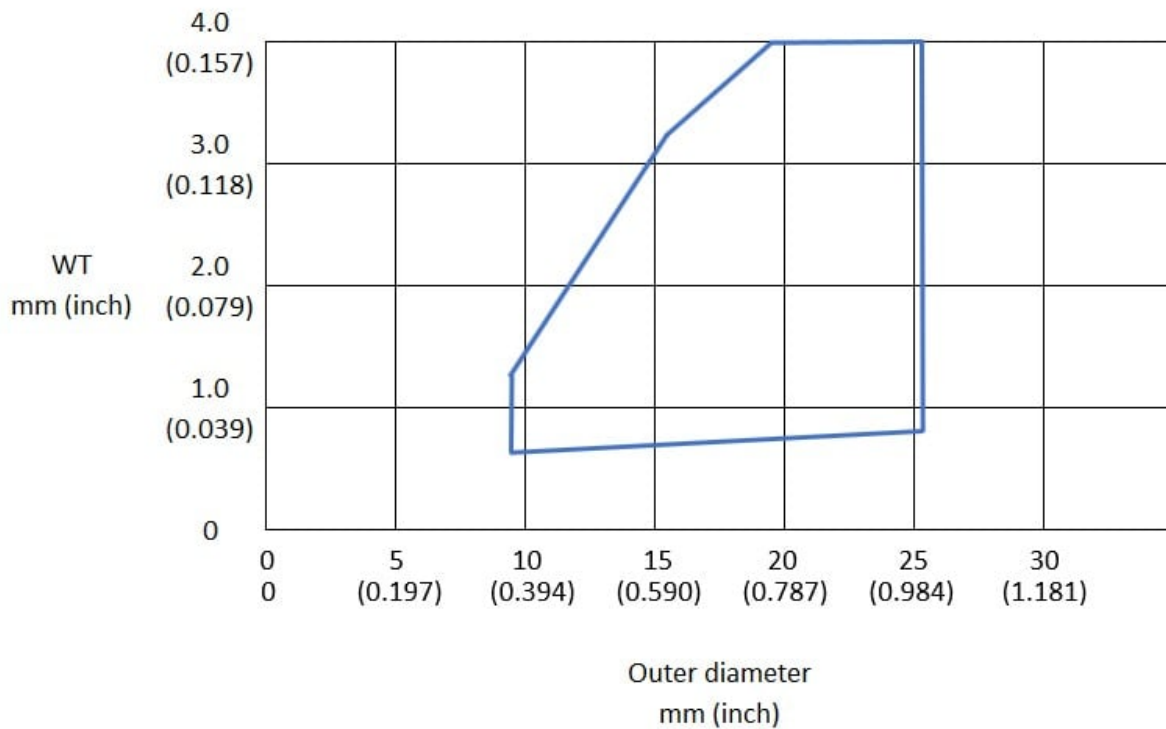


Figure 1 Principal size range for seamless tube and pipe.

OD	WT	L
9.50 - 24.4 mm	.07 - 4 mm	up to 17 m

Other dimensions can be quoted for special projects.

Mechanical properties

Tensile properties for Grade 2 titanium, as specified by ASTM B338, shown below:

	Ultimate Strength		Yield strength		Elongation 2"
	ksi	MPa	ksi	MPa	%
Min	50	345	40	275	20
Max			65	450	10
Typical data		530		380	39

A comparison of the yield strength of CP Titanium Grade 2 with other corrosion resistant tubing alloys is shown

in figure 3. *Min. 400 MPa for Titanium 2H

Hardness

92 HRB max.

Physical properties

Density 0.162 lbs/in³, 4.51 g/cm³

CP titanium tubing is lighter than comparable steel products, as its density is 45% less than ferrous alloys. Light weight and moderate strength give this product advantages where a strength-to-weight ratio better than stainless steel or CuNi alloys is required.

Melting point

3020°F (1660°C)

Beta transus

1675F (913C)

Thermal Conductivity

22 W/m°C (12.7 BTU) for RT
(Timet: 12,6 resp. 21,8. BPVC: 22-19.9 RT-200°C.)

Elastic modulus

The elastic modulus of Cp titanium, as shown below, is roughly one-half that of steel alloys.

	psi	GPa
Tension (E)	15.0 x 10 ⁶	103

Fatigue performance

Titanium, unlike many other materials, does not show a decrease in fatigue performance in the presence of

seawater. Both fatigue endurance limits and fatigue crack growth rates are the same whether tested in air or seawater.

Weldability

The weldability of CP titanium tubing is very good as long as the necessary precautions are taken. Due to the reactive nature of titanium inert gas shielding must be in place on both the OD and ID of the tubes. The material must be also free from any grease or oil contamination.

Manual or automatic TIG welding is regularly used to weld titanium tubing either with or without filler wire. A low heat input should be used to minimize the size of the heat-affected zone. No post-weld heat treatment is normally performed on titanium tubing.

Disclaimer: Recommendations are for guidance only, and the suitability of a material for a specific application can be confirmed only when we know the actual service conditions. Continuous development may necessitate changes in technical data without notice. This datasheet is only valid for Alleima materials.